



## **<sup>99m</sup>TECHNETIUM LABELLING OF ANTIBODIES USING METHYLPHOSPHONYLATED, BIFUNCTIONAL CYCLAMS AS CHELATORS**

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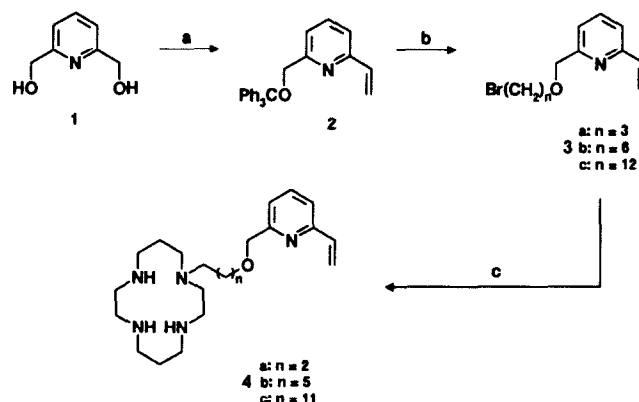
**Abstract:** Cyclam based bifunctional chelators are used for antibody labelling. The antibody conjugate **7**, containing the unmodified macrocycle, shows only slow complexation kinetics. The introduction of methylphosphonic acid groups into the cyclam (**9**) significantly enhances the complexation kinetics.

In today's medicine a variety of imaging techniques are well established. Most of them primarily show morphological changes. In contrast scintigraphy, a highly sensitive  $\gamma$ -radiation based technique, focuses on the visualisation of physiological changes. Immunoscintigraphy<sup>2)</sup> combines this principle with the high specificity of antibodies and uses it e. g. for tumor imaging. The most favourite  $\gamma$ -emitter for clinical applications is [<sup>99m</sup>Tc]technetium. It emits only  $\gamma$ -radiation, is available from a generator, has a low radiation energy and a short half live of only six hours.

In order to overcome the limitations of direct antibody labelling<sup>3)</sup> with <sup>99m</sup>Tc which is limited to certain classes of antibodies we have developed a generally applicable principle for the labelling of antibodies and other diagnostically relevant compounds<sup>4)</sup>.

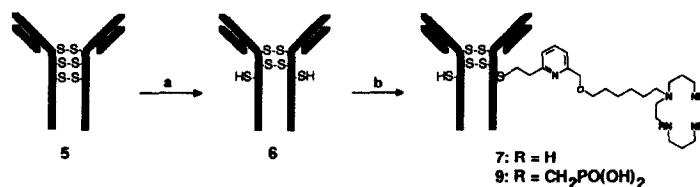
Polyazamacrocycles are well known chelators for cations<sup>5)</sup>. Especially 1,4,8,11-tetraazacyclotetradecan (cyclam) forms thermodynamically stable complexes with <sup>99m</sup>Tc which is required for medicinal applications. We decided to use a nitrogen atom of the cyclam to link the macrocycle to an antibody. In the past a variety of principles for linking macrocycles to proteins have been developed<sup>6)</sup>. We have chosen the vinylpyridin group which has been shown<sup>7)</sup> to selectively react with thiolgroups to covalently link the cyclam to the antibody. Free

thiol groups at the surface of antibodies can selectively be generated with 3-mercapto-1,2-propandiol<sup>8)</sup> without major alteration of their tertiary structure and biological activity.



**Reagents and conditions:** a: 1)  $\text{ClCPh}_3$ , Pyridine, DMAP, 1h RT, then 0,5h 60°C, 74%; 2)  $\text{MnO}_2$ , Dichloromethan; 6h RT, 77%; 3)  $\text{Ph}_3\text{PCH}_2\text{Br}/\text{NaNH}_2$ , THF, 4h RT, 75%; b: 1) HCl-gas, Dichloromethan, 0,5h RT, 95%; 2) Dibromopropane (a), Dibromohexan (b) or Dibromododecan (c), KOH, THF, 6h RT, 51-63%; c: 1,4,8,11-Tetraazacyclotetradecan, KOH, dioxan, 6h reflux, 72-78%.

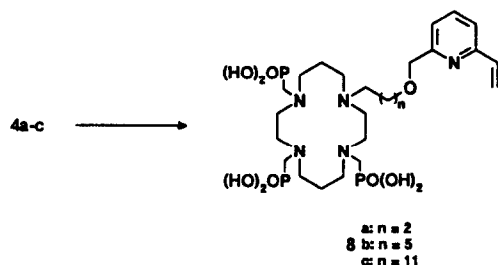
The synthesis of the linker starts with selective tritylation of 2,6-di-(hydroxymethyl)-pyridin **1** followed by manganese dioxide oxidation and a Wittig reaction generating the vinylpyridine **2**. Acidic cleavage of the trityl group and subsequent alkylation with dibromopropane, dibromohexane or dibromododecane led to **3a-c** having hydrophobic spacers with different length. Selective monoalkylation of cyclam with **3a-c** results in the desired bifunctional chelators **4a-c**.



**Reagents and conditions:** a: 3-Mercapto-1,2-propandiol; b: **4b** or **8b** (excess).

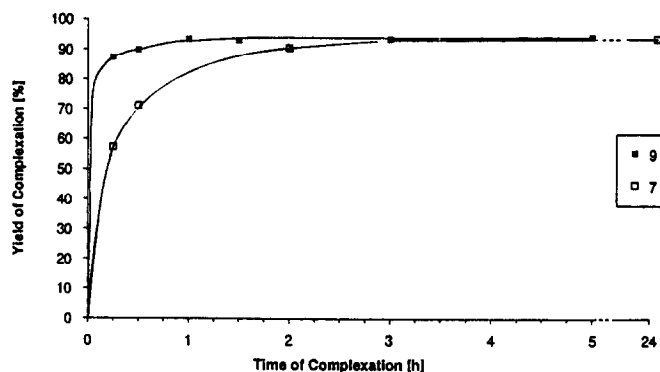
The human immunoglobulin Beriglobin<sup>®</sup> S<sup>9)</sup> **5**, the antibody used in this study, is treated with 3-mercapto-1,2-propandiol to generate **6** which has an average of 7 free thiol groups on its surface. These react with **4b**<sup>10)</sup> forming an average of 5.5 thioether linkages to the antibody conjugate **7**. Treatment of **7** with tin-(II)-citrate and sodium [<sup>99m</sup>Tc]pertechnetate-(VII) under

usual conditions<sup>11)</sup> yields 90% labelling of the antibody conjugate **7** after about 3 hours (figure 1,  $\square$ ). Complexation times around 3 hours are unacceptable considering the short half life of  $^{99m}\text{Tc}$  and a convenient clinical application.



**Reagents and conditions:** 1) 2N HCl, lyophilise; Diethylphosphite, Para-formaldehyde, Acetonitrile, 3Å-molecular sieves, HCl-gas, 7h reflux; 2) 5N HCl, 7h reflux.

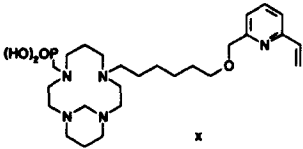
In order to optimise the chelation kinetics we introduced phosphonic acids which are well established  $^{99m}\text{Tc}$  chelators. E. g. methylenediphosphonic acids are used as diagnostics for bone cancer<sup>12)</sup>. The introduction of methylphosphonic acid units turned out to be quite tricky<sup>13)</sup>. Finally we have been able to obtain the desired trimethylphosphonic acids **6a-c** in 75% yield<sup>14)</sup>. The corresponding free acids **8a-c** are generated by acidic hydrolysis. The reaction of **8b** with the activated antibody **6** leads to the thioether **9** ( 4 thioethers in average per Ab).  $^{99m}\text{Tc}$  labelling yielded 95% complexation in only 30 minutes (figure 1,  $\blacksquare$ ).



**Figure 1:**  $^{99m}\text{Tc}$  labelling of the antibody conjugates **7** and **9**.

These results clearly demonstrate the advantage of the N-methylphosphonic acid residues. **8a-c** have proven to be superior bifunctional chelators with rapid complexation kinetics and high thermodynamic stability. With regards to antibody labelling the conjugate **9** fulfils the requirements for a potential clinical application. **8a-c** are currently under further investigation for the labelling of other antibodies and molecules of diagnostic value.

**Literature:**

- 1) Currently at: E. Merck, 64271 Darmstadt, Germany.
- 2) Magerstädt, M.; Hachmann, H. J.; Kuhlmann, L.; Seidel, L. *Chemiker-Zeitung* **114**, 1990, 123.
- 3) Hnatowich, D. J.; Mardirossian, G.; Ruschowski, M.; Fogarasi, M.; Virzi, F.; Winnard, P. *J. Nucl. Med.* **34**, 1993, 109.
- 4) For related approaches see for example: a) Franz, J.; Volkert, W. A.; Barefield, E. K.; Holmes, R. A. *Nucl. Med. Biol.* **14**, 1987, 569; b) Frizberg, A.R.; Abrams, P. G.; Baumier, P. L.; Kasina, S.; Morgan, A. C. Jr.; Rao, T. N.; Reno, J. M.; Sanderson, J. A.; Srinivasan, A.; Wilbur, D. S.; Vanderheyden, J. L. *Proc. Natl. Acad. Sci.* **85**, 1988, 4025; c) Baidoo, K. E.; Lever, S. Z. *Bioconjugate Chem.* **1**, 1990, 132; d) Lister-James, J.; Weber, R. W.; Boutin, R.; Nedelmann, M. A.; Dean, R. T. *J. Nucl. Med.* **30**, 1989, 793.
- 5) For a recent review see: Izatt, R. M.; Pawlak, K.; Bradshaw, J. S. *Chem. Rev.* **91**, 1991, 1721.
- 6) Parker, D. *Chem. Soc. Rev.* **19**, 1990, 271.
- 7) a) Hancock, R. D.; Martell, A. E. *Chem. Rev.* **89**, 1989, 1875; b) Morphy, J. R.; Parker, D.; Katakly, R.; Eaton, M. A. W.; Millican, A. T.; Alexander, R.; Harrison, A.; Walker, C. *J. Chem. Soc., Perkin Trans. 2* 1990, 573.
- 8) Zhang, Z. M.; Ballinger, J. R.; Sheldon, K.; Boxen, I. *Int. J. Nucl. Med. Biol.* **19**, 1992, 607.
- 9) The human immunoglobulin was used as a model for monoclonal antibodies because it was easily available in larger quantities.
- 10) **4b** was selected to be used with Beriglobin<sup>®</sup> S in this study.
- 11) For details of the chelation procedure see: Kuhlmann, L.; Pütter, D. EP 0498333.
- 12) Reiners, C. *Der Nuclearmediziner* **16**, 1993, 5.
- 13) There are several side reactions that can occur: Moisture in the reaction mixture results in the formation of large quantities of hydroxymethylphosphonic acid. A bigger problem represented the formation of the methylenbridged cyclam like for example **x**.  


**x**

Its formation can be avoided by using the hydrochlorides of **4a-c** as starting material. Furthermore the yields can be increased by adding catalytic amounts of HCl-gas.
- 14) For related compounds see: Broan, C. J.; Cole, E.; Jankowski, K. J.; Parker, D.; Pulukkody, K.; Boyce, B. A.; Beeley, N. R. A.; Millar, K.; Millican, A. T. *Synthesis* 1992, 63.

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